



education

Department of
Education
FREE STATE PROVINCE

CONTROL TEST / KONTROLETOETS

GRADE 10 / GRAAD 10

**PHYSICAL SCIENCES
FISIESE WETENSKAPPE**

MEMORANDUM

MARCH 2020 / MAART 2020

MARKS: 75 / PUNTE: 75

TIME: 1½ HOURS / TYD: 1½ UUR

**This memorandum consists of FIVE pages.
Hierdie memorandum bestaan uit VYF bladsye.**

QUESTION 1

- 1.1 D ✓✓
- 1.2 D ✓✓
- 1.3 B ✓✓
- 1.4 B ✓✓
- 1.5 B ✓✓
- 1.6 A ✓✓
- 1.7 D ✓✓
- 1.8 B ✓✓

[16]

QUESTION 2 / VRAAG 2

- 2.1 A pure substance consisting of two or more different elements. ✓✓
'n Suiwer stof bestaande uit twee of meer verskillende elemente. (2)
 - 2.2 A mixture of uniform composition and in which all components are in the same phase. ✓✓
'n Mengsel van uniforme samestelling en waarin alle komponente in dieselfde fase is. (2)
 - 2.3.1 C ✓ (1)
 - 2.3.2 A✓ & C ✓ (2)
 - 2.3.3 D✓ & E✓ & B (any two/enige twee) (2)
 - 2.3.4 B ✓ (1)
 - 2.3.5 A or/of C ✓ (1)
 - 2.3.6 E ✓ (1)
 - 2.3.7 D ✓ (1)
- [13]

QUESTION 3 / VRAAG 3

3.1 The temperature at which a solid, given sufficient heat, becomes a liquid. ✓✓

Die temperatuur waar 'n vaste stof, indien dit voldoende hitte verkry, 'n vloeistof word. (2)

3.2 Temperature/Temperatuur ✓ (1)

3.3 Boil/Kook ✓ (1)

3.4 Liquid/Vloeistof ✓ (1)

3.5 F ✓ (1)

[6]

QUESTION 4 / VRAAG 4

4.1 The number of protons and neutrons in an atom. ✓✓

Die aantal protone en neutrone in 'n atoom. (2)

4.2 Atoms of the same element having the same number of protons, but different numbers of neutrons. ✓✓

Atome van dieselfde element wat dieselfde aantal protone, maar verskillende aantal neutrone, het. (2)

4.3.1 17 ✓ (1)

4.3.2 17 ✓ (1)

4.3.3 20 ✓ (1)

4.4 Ammonium dichromate / Ammoniumdichromaat ✓ (1)

4.5 PbCl_2 ✓ (1)

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QUESTION 5 / VRAAG 5

5.1 The sharing of electrons between atoms to form molecules. ✓✓

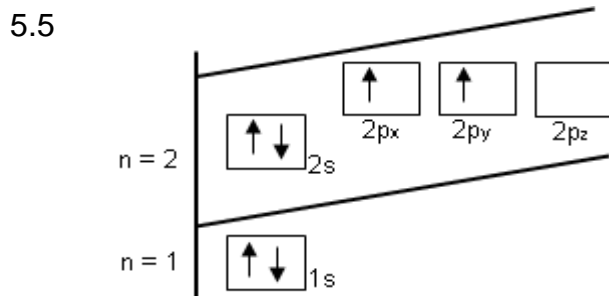
Die deel van elektrone tussen atome om molekule te vorm. (2)

5.2.1  (1)

5.2.2  (1)

5.3 Pauli ✓ (1)

5.4 4 ✓ (1)



Criteria for Aufbau diagram / Kriteria vir Aufbaudiagram:	Mark/Punt
Drawn for two main energy levels. <i>Geteken vir eerste twee hoofenergievlakke.</i>	✓
First four electrons represented as pairs with opposite spin. <i>Eerste vier elektrone getoon as pare met teenoorgestelde spin.</i>	✓
Last two electrons represented as unpaired. <i>Laaste twee elektrone as ongepaard aangedui.</i>	✓

(3)

5.6.1 $1s^1$ ✓ (1)

5.6.2 CH_4 ✓ (1)

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QUESTION 6 / VRAAG 6

6.1.1 A&C OR/OF any two of/enige twee van E, B & D ✓ (1)

6.1.2 AC or/of BD or/of EB ✓ (1)

6.2.1 0,75 m ✓ (1)

6.2.2 5 m ✓ (1)

6.3 The number of waves per second ✓✓

Die aantal golwe per sekonde (2)

6.4.1 $f = \frac{n}{\Delta t} \checkmark = \frac{2}{3} \checkmark = 0,67 \text{ Hz} \checkmark$ OR/OF $f = \frac{1}{T} \checkmark = \frac{1}{1,5} \checkmark = 0,67 \text{ Hz} \checkmark$ (3)

6.4.2 **POSITIVE MARKING FROM 6.4.1. / POSITIEWE NASIEN VANAF 6.4.1.**

$$T = \frac{1}{f} \checkmark = \frac{1}{0,67} \checkmark = 1,49 \text{ s} \checkmark (1,50 \text{ s})$$

OR/OF

$$T = 1,5 \text{ s} (\checkmark\checkmark)$$

It takes 3 s to complete two waves. ✓

Dit neem 3 s om twee golwe te voltooi.

(3)
[12]

QUESTION 7 / VRAAG 7

7.1.1 $T = \frac{1}{f} = \frac{1}{200} \checkmark = 0,005 \text{ s} \checkmark$ (2)

7.1.2 **POSITIVE MARKING FROM 7.1.1. / POSITIEWE NASIEN VANAF 7.1.1.**

$$\begin{array}{ll} v = \lambda f \checkmark & \bar{v} = \frac{\Delta x}{\Delta t} \checkmark \\ 320 = \lambda(200) \checkmark & \text{OR/OF} \\ \lambda = 1,6 \text{ m} \checkmark & 320 = \frac{\lambda}{0,005} \checkmark \\ & \lambda = 1,6 \text{ m} \checkmark \end{array} \quad (3)$$

$$\begin{array}{ll} 7.2 & \bar{v} = \frac{\Delta x}{\Delta t} \\ & \checkmark 340 = \frac{400}{\Delta t} \checkmark \\ & \Delta t = 1,18 \text{ s} \checkmark \end{array} \quad \text{OR/OF} \quad \begin{array}{ll} & \bar{v} = \frac{\Delta x}{\Delta t} \\ & \checkmark 340 = \frac{200}{\Delta t} \checkmark \\ & \Delta t = 0,588 \text{ s} \\ & \Delta t_t = 2 \times 0,588 \\ & = 1,18 \text{ s} \checkmark \end{array} \quad (3) \quad [8]$$

GRAND TOTAL / GROOTTOTAAL: 75